

**Data collection**

Enraf–Nonius CAD-4 diffractometer  
 $\omega/2\theta$  scans  
 Absorption correction: empirical *via*  $\psi$  scans (Fair, 1990)  
 $T_{\min} = 0.973$ ,  $T_{\max} = 0.975$   
 2542 measured reflections  
 2352 independent reflections

1974 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.008$   
 $\theta_{\max} = 26.3^\circ$   
 $h = -8 \rightarrow 8$   
 $k = -9 \rightarrow 0$   
 $l = -14 \rightarrow 14$   
 3 standard reflections  
 frequency: 120 min  
 intensity decay: 0.7%

**Refinement**

Refinement on  $F^2$   
 $R = 0.033$   
 $wR = 0.043$   
 $S = 1.11$   
 1974 reflections  
 172 parameters  
 H atoms: see below  
 $w = \sigma^2(F) + 0.0004F^2 + 0.50$

$(\Delta/\sigma)_{\max} = 0.0002$   
 $\Delta\rho_{\max} = 0.25 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.20 \text{ e } \text{\AA}^{-3}$   
 Extinction correction: none  
 Scattering factors from *International Tables for X-ray Crystallography* (Vol. IV)

Table 1. Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

C1—C4	1.745 (2)	O2—C7	1.224 (3)
S—C8	1.792 (2)	N—C6	1.412 (2)
S—C9	1.766 (2)	N—C7	1.335 (2)
O1—C1	1.361 (3)	C7—C8	1.521 (3)
C8—S—C9	103.2 (1)	N—C6—C5	124.4 (2)
C6—N—C7	128.8 (2)	O2—C7—N	124.1 (2)
O1—C1—C2	123.9 (2)	O2—C7—C8	119.5 (2)
O1—C1—C6	116.4 (1)	N—C7—C8	116.4 (2)
C1—C4—C3	119.5 (2)	S—C8—C7	118.8 (1)
C1—C4—C5	118.6 (2)	S—C9—C10	123.5 (2)
N—C6—C5	115.6 (2)	S—C9—C14	117.5 (2)
C9—S—C8—C7	80.3 (2)	O2—C7—C8—S	-160.3 (2)
C6—N—C7—C8	178.1 (2)		

Table 2. Hydrogen-bonding geometry ( $\text{\AA}$ ,  $^\circ$ )

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
O1—H01 $\cdots$ O2 <sup>1</sup>	0.94	1.74	2.683 (4)	176
C2—H2 $\cdots$ O2 <sup>1</sup>	0.95	2.63	3.304 (5)	128

Symmetry code: (i)  $x - 1, y, z$ .

All non-H atoms were refined with anisotropic displacement parameters. The H atoms, except H01, H1, H81 and H82, were placed geometrically 0.95  $\text{\AA}$  from their corresponding C atoms, with  $U_{\text{iso}}(\text{H}) = 1.3U_{\text{eq}}(\text{C})$ . The H01, H1, H81 and H82 atoms were taken from a difference Fourier map. A riding model was used for all H atoms.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1993). Data reduction: *MolEN* (Fair, 1990). Program(s) used to solve structure: *SIMPEL* in *MolEN*. Program(s) used to refine structure: *LSFM* in *MolEN*. Molecular graphics: *MolEN* version of *ORTEP*. Software used to prepare material for publication: *MolEN*.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: MU1319). Services for accessing these data are described at the back of the journal.

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**4-Nitro-1-(trimethylsilylethynyl)benzene**

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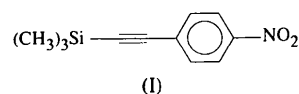
(Received 25 April 1997; accepted 22 September 1997)

**Abstract**

The title molecule,  $\text{C}_{11}\text{H}_{13}\text{NO}_2\text{Si}$ , lies on a mirror plane, with only one methyl group lying out of plane. The  $\text{C}\equiv\text{C}$  triple bond has a length of 1.199 (4)  $\text{\AA}$ . Bond angles  $\text{Si}-\text{C}\equiv\text{C}$  and  $\text{C}\equiv\text{C}-\text{C}(\text{Ar})$  are 177.9 (3) and 178.0 (3)°, respectively. The  $\text{Si}-\text{C}_{\text{sp}^3}$  bond lengths are 1.831 (4) and 1.838 (3)  $\text{\AA}$ , while the  $\text{Si}-\text{C}_{\text{sp}}$  distance is 1.839 (3)  $\text{\AA}$ .

**Comment**

The title compound, (I), was prepared as part of a structural study involving substituted silylethynylbenzene derivatives. The molecule lies on a mirror plane, with



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only the C10 atom lying out of plane. The acetylenic bond distance [C1—C2 1.199 (4) Å] is in agreement with the average value for C<sub>sp</sub>≡C<sub>sp</sub>—C<sub>sp</sub><sup>2</sup>(Ar) reported by Allen *et al.* (1987), and is also in agreement with those of related compounds, *i.e.* 2,3,5,6-tetrakis[1-ethynyl-2-(trimethylsilyl)]pyridine [1.190 (4) and 1.191 (4) Å; Garcia *et al.*, 1997], 2,3,5,6-tetrakis[1-ethynyl-2-(trimethylsilyl)]pyrazine [1.198 (5) and 1.199 (6) Å; Garcia *et al.*, 1997], 2,3,4,5-tetrakis[1-ethynyl-2-(trimethylsilyl)]thiophene [1.190 (4) and 1.190 (4) Å; Garcia *et al.*, 1997], 3,4-dibromo-2,5-bis[1-ethynyl-2-(trimethylsilyl)]-*N*-methylpyrrole [1.189 (5) and 1.194 (4) Å; Garcia *et al.*, 1997], 2-[(2,6-dihydroxyphenyl)ethynyl]benzoic acid [1.195 (2) Å; Huang *et al.*, 1992], methyl 2-[(2,6-dimethoxyphenyl)ethynyl]benzoate [1.195 (2) Å; Huang *et al.*, 1991], methyl 2-[(2,6-dimethoxyphenyl)ethynyl]-3-methoxybenzoate [1.195 (2) Å; Evans *et al.*, 1990], 3,3-diphenyl-1-propyn-3-ol [1.169 (2) Å; Garcia *et al.*, 1995], phenyl 2-(trimethylsilylethynyl)phenyl methanone and 2-(trimethylsilylethynyl)benzaldehyde [1.199 (3) and 1.192 (3) Å, respectively; Garcia *et al.*, 1996], 3-[4-(3-cyanophenylethynyl)phenyl]-5-methoxymethyl-2-oxazolidinone [1.190 (20) Å; Durant *et al.*, 1982] and bis(*p*-nitrophenyl)butadiyne [1.195 (4) Å; Mayerle *et al.*, 1979].

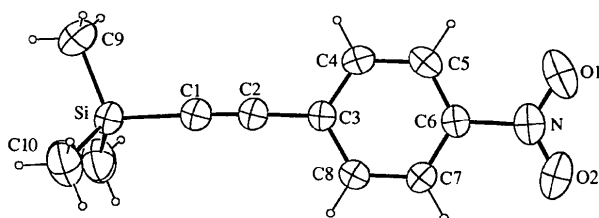


Fig. 1. The title molecule with ellipsoids at the 40% probability level and H atoms represented with arbitrary radii.

The packing is illustrated in Fig. 2, in which the herring-bone pattern is evident. The molecules lie on a mirror plane allowing them to stack in the **b** direction with parallel aromatic rings and a spacing of **b**/2 = 3.6553 (3) Å. The rings are offset from exact overlap, however, with their centroids separated by 4.179 Å. These centroids form zigzag chains in the **b** direction, forming angles of 122.0°.

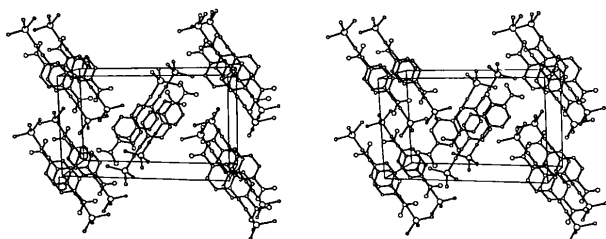


Fig. 2. Stereoview of the unit cell, viewed slightly oblique to the **b** direction, with **c** horizontal. H atoms are not shown.

## Experimental

The title compound was prepared by a palladium(II)-catalyzed coupling (Austin *et al.*, 1980) of trimethylsilylacetylene with 4-nitroiodobenzene.

### Crystal data

C<sub>11</sub>H<sub>13</sub>NO<sub>2</sub>Si  
*M<sub>r</sub>* = 219.32  
 Orthorhombic  
*Pnma*  
*a* = 10.3139 (9) Å  
*b* = 7.3105 (5) Å  
*c* = 16.735 (1) Å  
*V* = 1261.8 (3) Å<sup>3</sup>  
*Z* = 4  
*D<sub>x</sub>* = 1.154 Mg m<sup>-3</sup>  
*D<sub>m</sub>* not measured

Cu *Kα* radiation  
 $\lambda$  = 1.54184 Å  
 Cell parameters from 25 reflections  
 $\theta$  = 9–28°  
 $\mu$  = 1.50 mm<sup>-1</sup>  
*T* = 298 K  
 Needle  
 0.48 × 0.22 × 0.13 mm  
 Yellow

### Data collection

Enraf–Nonius CAD-4 diffractometer  
 $\theta/2\theta$  scans  
 Absorption correction:  $\psi$  scans (North *et al.*, 1968)  
 $T_{\min}$  = 0.773,  $T_{\max}$  = 0.823  
 2606 measured reflections  
 1405 independent reflections

1027 reflections with  $I > \sigma(I)$   
 $R_{\text{int}}$  = 0.019  
 $\theta_{\text{max}}$  = 75°  
 $h$  = 0 → 12  
 $k$  = 0 → 9  
 $l$  = -9 → 20  
 3 standard reflections  
 frequency: 120 min  
 intensity decay: 1.6%

### Refinement

Refinement on *F*  
 $R$  = 0.046  
 $wR$  = 0.051  
 $S$  = 2.17  
 1027 reflections  
 89 parameters  
 H atoms calculated, with  $U = 1.3U$  of the bonded atom  
 $w = 4F_o^2 / [\sigma^2(F_o^2) + 0.0004F_o^4]$

$(\Delta/\sigma)_{\text{max}}$  = 0.07  
 $\Delta\rho_{\text{max}}$  = 0.15 e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}}$  = -0.16 e Å<sup>-3</sup>  
 Extinction correction: isotropic (Zachariasen, 1963)  
 Extinction coefficient:  $4.6(3) \times 10^{-6}$   
 Scattering factors from *International Tables for X-ray Crystallography* (Vol. IV)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å<sup>2</sup>)

$$U_{\text{eq}} = (1/3) \sum_i \sum_j U^{ij} a_i^* a_j^* \mathbf{a}_i \cdot \mathbf{a}_j$$

	<i>x</i>	<i>y</i>	<i>z</i>	<i>U<sub>eq</sub></i>
Si	0.47230 (8)	1/4	-0.14033 (5)	0.0794 (3)
O1	-0.3165 (2)	1/4	0.1330 (2)	0.155 (1)
O2	-0.1936 (3)	1/4	0.2321 (2)	0.198 (2)
N	-0.2104 (2)	1/4	0.1627 (2)	0.106 (1)
C1	0.3218 (3)	1/4	-0.0814 (2)	0.082 (1)
C2	0.2260 (3)	1/4	-0.0407 (2)	0.079 (1)
C3	0.1143 (2)	1/4	0.0103 (1)	0.0656 (8)
C4	-0.0108 (3)	1/4	-0.0206 (2)	0.084 (1)
C5	-0.1167 (3)	1/4	0.0286 (2)	0.086 (1)
C6	-0.0966 (3)	1/4	0.1094 (2)	0.0718 (9)
C7	0.0235 (3)	1/4	0.1419 (2)	0.079 (1)
C8	0.1292 (3)	1/4	0.0926 (2)	0.075 (1)
C9	0.4288 (4)	1/4	-0.2464 (2)	0.133 (2)
C10	0.5687 (2)	0.0470 (4)	-0.1145 (2)	0.122 (1)

Table 2. Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

O1—N	1.201 (4)	C1—C2	1.199 (4)
O2—N	1.175 (4)		
O1—N—O2	122.9 (3)	Si—C1—C2	177.9 (3)

Data collection: *CAD-4 Operations Manual* (Enraf–Nonius, 1977). Cell refinement: *CAD-4 Operations Manual*. Data reduction: *PROCESS* in *MolEN* (Fair, 1990). Program(s) used to solve structure: *MULTAN80* (Main *et al.*, 1980). Program(s) used to refine structure: *LSFM* in *MolEN*. Molecular graphics: *ORTEPII* (Johnson, 1976). Software used to prepare material for publication: *CIF IN* in *MolEN*.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: BK1347). Services for accessing these data are described at the back of the journal.

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## Structure and Photochemistry of Four Adamantylacetophenones

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## Abstract

The photochemistry and crystal structures of four  $\alpha$ -adamantylacetophenones have been studied, namely, 1-(4-fluorophenyl)-2-(tricyclo[3.3.1.1<sup>3,7</sup>]dec-1-yl)ethanone [(1a), C<sub>18</sub>H<sub>21</sub>FO], 1-(4-cyanophenyl)-2-(tricyclo[3.3.1.1<sup>3,7</sup>]dec-1-yl)ethanone [(1b), C<sub>19</sub>H<sub>21</sub>NO], 1-(4-cyanophenyl)-2-(3-methyltricyclo[3.3.1.1<sup>3,7</sup>]dec-1-yl)ethanone [(1c), C<sub>20</sub>H<sub>23</sub>NO] and 1-(4-carboxyphenyl)-2-(3-methyltricyclo[3.3.1.1<sup>3,7</sup>]dec-1-yl)ethanone [(1d), C<sub>20</sub>H<sub>24</sub>O<sub>3</sub>]. All four molecules adopt fairly similar conformations, with short  $\gamma\text{H}\cdots\text{O}$  contacts permitting hydrogen abstraction in the photochemical reactions. However, three of the compounds, (1a), (1b) and (1d), are photostable in the solid state and possible reasons for this unreactivity are considered.

## Comment

Study of the photochemical Norrish type II  $\gamma$ -hydrogen-abstraction reaction of  $\alpha$ -adamantylacetophenones is simplified by the fact that products formed by cleavage of the intermediate 1,4-biradicals are not possible, since they would include the highly strained adamantene (Evans & Trotter, 1989*a,b*; Jones *et al.*, 1994). For reactants of type (1) ( $R = \text{H}$ ), the favoured products are two cyclobutanols, with *cis*-OH, (2), or *trans*-OH substituents (relative to the ring junction H atom). The major photoproduct is usually the *trans*-OH isomer, which is presumably more stable than the *cis*-OH isomer, since the latter exhibits greater steric repulsions between a pseudo-axial aryl substituent on the folded four-membered ring and the adamantyl group. Production of the *trans*-OH isomer involves rotation of the adamantyl group by about 90° from the observed solid-state molecular conformation of the adamantylacetophenones; since even the solid-state photoreactions yield the *trans*-OH isomer as the major product, rotation of the approximately spherical adamantyl group appears to be a preferred (energetically favourable) reaction pathway.